

1-(2,4-Dichlorophenyl)-3-(3,4-dimethoxyphenyl)prop-2-en-1-one

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Key indicators

Single-crystal X-ray study
 $T = 100\text{ K}$
Mean $\sigma(\text{C}-\text{C}) = 0.004\text{ \AA}$
 R factor = 0.051
 wR factor = 0.153
Data-to-parameter ratio = 16.8

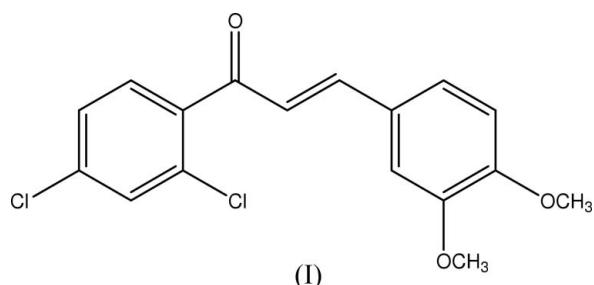
For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

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In the title compound, $\text{C}_{17}\text{H}_{14}\text{Cl}_2\text{O}_3$, the dihedral angle between the benzene rings is $45.42(15)^\circ$. In the crystal structure, molecules form layers parallel to the bc plane, and these are stabilized by intermolecular $\text{C}-\text{H}\cdots\text{O}$ hydrogen bonds and by $\pi-\pi$ interactions between the benzene rings with a centroid–centroid distance of 3.848 \AA .

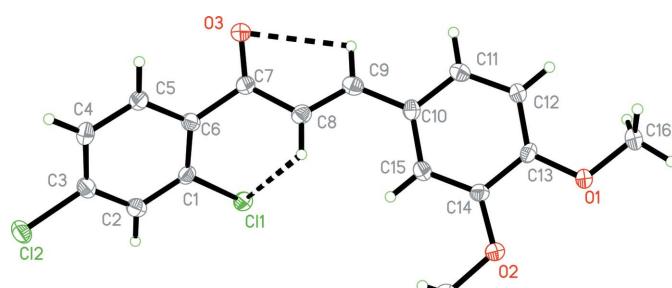
Comment

Chalcones show an impressive array of pharmaceutical properties, such as antiprotozoal (Nielsen *et al.*, 1998; Li *et al.*, 1995, Liu *et al.*, 2001), anti-inflammatory (Hsieh *et al.*, 1998) and nitric oxide inhibition activities (Rojas *et al.*, 2002). Recently, it has been noted that derivatives of chalcones exhibit extremely high and fast non-linearity (Fichou *et al.*, 1988; Kitaoka *et al.*, 1990; Uchida *et al.*, 1998; Goto *et al.*, 1991; Patil *et al.*, 2006*a,b*; Zhang *et al.*, 1990; Zhao *et al.*, 2000). We and others have undertaken a number of theoretical and structural studies of such compounds (Ng *et al.*, 2006; Patil *et al.*, 2006*a,b*; Teh *et al.*, 2006; Radha Krishna *et al.*, 2005; Sathiya Moorthi *et al.*, 2005; Uchida *et al.*, 1995), and we report here the structure of the title compound (I) (Fig. 1). Crystals of (I) do not exhibit second-order non-linear optical properties as they crystallize in a centrosymmetric space group.

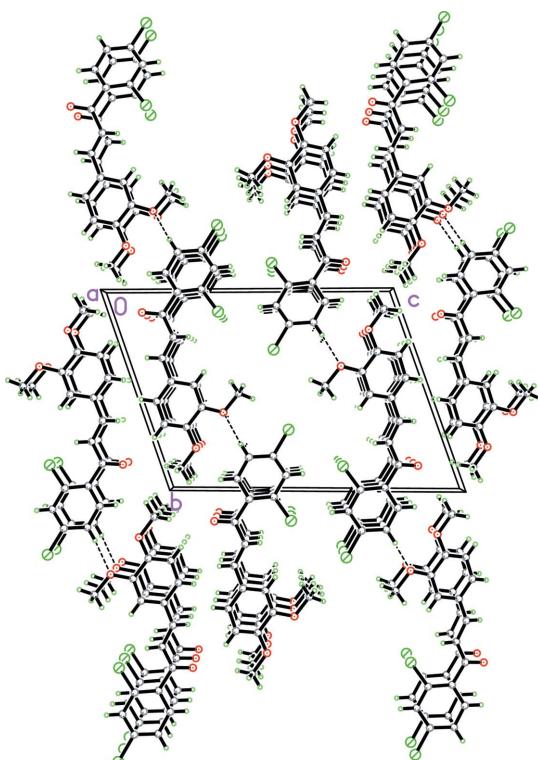


Bond lengths and angles for (I) are within normal ranges (Allen *et al.*, 1987) and are comparable with those in related structures (Ng *et al.*, 2006; Patil *et al.*, 2006*a,b*; Teh *et al.*, 2006; Radha Krishna *et al.*, 2005; Sathiya Moorthi *et al.*, 2005). The dihedral angle between the benzene rings is $45.42(15)^\circ$. The enone group makes dihedral angles of $46.15(8)$ and $1.22(14)^\circ$ with the C1–C6 and C10–C15 rings, respectively. The two methoxy groups attached at C13 and C14 are almost coplanar with the C10–C15 benzene ring, with C16–O1–C13–C12 and C17–O2–C14–C15 torsion angles of $-1.1(4)$ and $0.5(4)^\circ$, respectively.

Two intramolecular hydrogen bonds, C9–H9A \cdots O3 and C8–H8A \cdots Cl1 (Table 1), generate S(5) and S(6) ring motifs, respectively, in the molecule (Bernstein *et al.*, 1995). The

**Figure 1**

The asymmetric unit of (I), showing 50% probability displacement ellipsoids and the atom-numbering scheme. Hydrogen bonds are shown as dashed lines.

**Figure 2**

The crystal packing of (I), viewed down the a axis. Hydrogen bonds are shown as dashed lines.

crystal structure consists of layers of molecules parallel to the bc plane and stabilized by intermolecular $C_4-H_4A\cdots O_2^+$ hydrogen bonds along the b axis (Table 1). The crystal structure is also stabilized by $\pi-\pi$ interactions between the benzene rings along the a axis, with a centroid-centroid distance of 3.848 Å.

Experimental

Compound (I) was obtained by the condensation of 3,4-dimethoxybenzaldehyde (0.01 mol) with 2,4-dichloroacetophenone (0.01 mol) in ethanol (60 ml) in the presence of NaOH (5 ml, 20%). After stirring for 4 h, the contents of the flask were poured into ice-cold water (250 ml), and the resulting crude solid was collected by filtration. The compound was dried and purified by repeated recrystallization from

acetone. Crystals of (I), suitable for a single-crystal diffraction study, were grown by slow evaporation of an acetone solution.

Crystal data

$C_{17}H_{14}Cl_2O_3$	$V = 743.36 (6)$ Å 3
$M_r = 337.18$	$Z = 2$
Triclinic, $P\bar{1}$	$D_x = 1.506$ Mg m $^{-3}$
$a = 3.8479 (2)$ Å	Mo $K\alpha$ radiation
$b = 12.2341 (6)$ Å	$\mu = 0.45$ mm $^{-1}$
$c = 16.4845 (7)$ Å	$T = 100.0 (1)$ K
$\alpha = 70.487 (3)$ °	Plate, yellow
$\beta = 84.170 (4)$ °	$0.35 \times 0.15 \times 0.04$ mm
$\gamma = 86.707 (3)$ °	

Data collection

Bruker SMART APEX2 CCD area-detector diffractometer	12829 measured reflections
ω scans	3372 independent reflections
Absorption correction: multi-scan (<i>SADABS</i> ; Bruker, 2005)	2434 reflections with $I > 2\sigma(I)$
$R_{\text{int}} = 0.082$	
$T_{\min} = 0.814$, $T_{\max} = 0.981$	$\theta_{\max} = 27.5$ °

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.051$	$w = 1/[\sigma^2(F_o^2) + 0.6876P]$
$wR(F^2) = 0.153$	where $P = (F_o^2 + 2F_c^2)/3$
$S = 1.05$	$(\Delta/\sigma)_{\max} < 0.001$
3372 reflections	$\Delta\rho_{\max} = 0.34$ e Å $^{-3}$
201 parameters	$\Delta\rho_{\min} = -0.39$ e Å $^{-3}$

Table 1
Hydrogen-bond geometry (Å, °).

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
$C4-H4A\cdots O2^i$	0.93	2.57	3.475 (4)	165
$C8-H8A\cdots Cl1$	0.93	2.78	3.139 (3)	103
$C9-H9A\cdots O3$	0.93	2.55	2.866 (4)	100

Symmetry code: (i) $x, y - 1, z$.

H atoms were placed in calculated positions, with C–H distances in the range 0.93–0.96 Å. $U_{\text{iso}}(\text{H})$ values were set equal to $1.5U_{\text{eq}}$ of the carrier atom for methyl H atoms and $1.2U_{\text{eq}}$ for the remaining H atoms.

Data collection: *APEX2* (Bruker, 2005); cell refinement: *APEX2*; data reduction: *SAINT* (Bruker, 2005); program(s) used to solve structure: *SHELXTL* (Sheldrick, 1998); program(s) used to refine structure: *SHELXTL*; molecular graphics: *SHELXTL*; software used to prepare material for publication: *SHELXTL*, *PARST* (Nardelli, 1995) and *PLATON* (Spek, 2003).

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